

**Handout 8**  
Course equations

**Interactions**

$$u(r) = 4\epsilon \left[ \left( \frac{r}{\sigma} \right)^{-12} - \left( \frac{r}{\sigma} \right)^{-6} \right]$$

Lennard-Jones

$$u(r) = \frac{q_1 q_2}{4\pi\epsilon_0 r}$$

Coulomb's law

**Thermodynamics**

$$S = k_B \ln \Omega$$

$$\Delta S = \int \frac{\delta Q_{rev}}{T}$$

$$G \equiv U - TS + PV$$

$$H \equiv U + PV$$

$$A \equiv U - TS$$

$$C_V \equiv \left( \frac{dU}{dT} \right)_V = T \left( \frac{dS}{dT} \right)_V \quad C_P \equiv \left( \frac{dH}{dT} \right)_P = T \left( \frac{dS}{dT} \right)_P$$

$$\wp_m \propto e^{-\frac{U_m}{k_B T}}$$

$$\wp(\text{state}) \propto \exp \left[ -\frac{G(\text{state})}{RT} \right]$$

$$\Delta G(T) = \Delta H_0 \left( 1 - \frac{T}{T_0} \right) + \Delta C_p \left[ T - T_0 - T \ln \left( \frac{T}{T_0} \right) \right] \quad (\text{constant } C_p \text{ model})$$

**Chemical kinetics**

$$k = k_0 e^{-\frac{\Delta G^\ddagger}{RT}}$$

$$\Delta G_{rxn} = \Delta G_{rxn}^\circ + RT \ln \frac{[B]}{[A]}$$

$$K_{eq} \equiv \exp \left( -\frac{\Delta G_{rxn}^\circ}{RT} \right)$$

$$K_{eq} = \frac{k_f}{k_r}$$

**Protein folding**

$$\frac{[F]}{[U]} = K = e^{-\frac{\Delta G_{fold}^\circ}{RT}}$$

$$[F] = \frac{k_f [1 - e^{-(k_f + k_u)t}]}{k_f + k_u} [U]_0$$

## Ligand binding

$$\frac{[C]}{[P][L]} = K_A = \frac{1}{K_D} = \exp\left(-\frac{\Delta G_{\text{bind}}^0}{RT}\right)$$

$$\ln \frac{K_D(T_2)}{K_D(T_1)} = \frac{\Delta H(T_1)}{R} \left(\frac{1}{T_2} - \frac{1}{T_1}\right) + \frac{\Delta C_P}{R} \left[1 - \frac{T_1}{T_2} - \ln\left(\frac{T_2}{T_1}\right)\right] \quad (\text{constant } C_p \text{ model})$$

$$\frac{[P]}{[P]_0} = \frac{1}{1 + K_{eq}[L]^n} \quad (\text{Hill equation})$$

## Enzyme kinetics

$$\frac{d[P]}{dt} = v = \frac{v_{\text{max}}[S]}{[S] + K_M} \quad v_{\text{max}} \equiv k_2[E]_0 \quad K_M \equiv \frac{k_{-1} + k_2}{k_1} \quad (\text{Michaelis-Menten})$$

mechanism	$v_{\text{max,app}}$	$K_{M,app}$	special cases
competitive inhibition	$v_{\text{max}}$	$K_M \left(1 + \frac{[I]}{K_I}\right)$	<i>product inhibition:</i> $v$ decreases with time
non-competitive inhibition	$v_{\text{max}} \left(1 + \frac{[I]}{K_I'}\right)^{-1}$	$K_M \left(1 + \frac{[I]}{K_I'}\right)^{-1}$	<i>substrate inhibition:</i> $v$ decreases more than expected with increasing $[S]$
mixed inhibition	$v_{\text{max}} \left(1 + \frac{[I]}{K_I'}\right)^{-1}$	$K_M \left(1 + \frac{[I]}{K_I}\right) \left(1 + \frac{[I]}{K_I'}\right)^{-1}$	
pH-sensitive	$v_{\text{max}}$	$K_{M,app} \equiv K_M \left(1 + \frac{K_3}{[H^+]} + \frac{[H^+]}{K_4}\right)$	

## Membranes

$$[M]_0 = [M] \left[1 + \left(\frac{2[M]}{c^*}\right)^{n-1}\right] \quad K_{eq} = \frac{1}{n} \left(\frac{2}{c^*}\right)^{n-1} \quad (\text{critical micelle concentration})$$

$$\Pi = RT\Delta c \quad (\text{osmotic pressure})$$

$$\Delta V = -\frac{k_B T}{qe} \ln\left(\frac{c_i}{c_e}\right) \quad (\text{Nernst equation})$$

$$\approx \mp (60 \text{ mV}) \log_{10}\left(\frac{c_i}{c_e}\right) \quad \text{at } T = 300 \text{ K for } \pm 1 \text{ charged ions}$$

## Transport

$$J = P(c_1 - c_2) \quad P = \frac{DH}{l}$$

$$D = \frac{k_B T}{6\pi\mu R} \quad (\text{Stokes-Einstein})$$

$$\langle r^2 \rangle = 6Dt \quad (\text{mean squared displacement})$$

$$[P]_1 = \frac{v_{in}}{k_1} (1 - e^{-k_1 t}) \quad [P]_2 = \frac{v_{in}}{k_2} \left(1 + \frac{k_1 e^{-k_2 t} - k_2 e^{-k_1 t}}{k_2 - k_1}\right) \quad (\text{two-compartment model})$$